

# METHOD FOR THE TREATMENT OF COAL

## BACKGROUND OF THE INVENTION

5

### 1. Field of the Invention

The present invention relates to a method for the conversion of coal to recover energy value from the coal while minimizing waste by-products from the conversion process. The method can advantageously be utilized to convert very low-grade coal to products having high-energy value without emitting high levels of pollutants into the atmosphere.

### 2. Description of Related Art

Recently, the United States and other countries have experienced a shortage of natural gas and as a result, natural gas prices for consumers have increased substantially. Accordingly, there is a pressing need for economic methods for the manufacture of a high-value heating gas that can be used in place of natural gas. Natural gas has in composition that includes about 85 percent methane ( $\text{CH}_4$ ), about 10 percent ethane ( $\text{C}_2\text{H}_6$ ) and the balance including propane ( $\text{C}_3\text{H}_8$ ), butane ( $\text{C}_4\text{H}_{10}$ ) and nitrogen ( $\text{N}_2$ ). Methane, the primary component of natural gas, has a heating value of about 51,623 Btu/lb.

In addition to being supplied directly to consumers, natural gas and other high BTU gas compositions can be utilized on-site to produce electricity. Further, there are many natural resources that are utilized to produce energy in addition to natural gas. For example, coal can be burned in conventional boilers to generate steam, which can be used to generate electricity. Many regions such as the United States, China and India possess vast reserves of coal that are not presently being exploited for current energy demands due to the high economic and ecological costs related to the use of the low-grade coal. Further, some of the coal deposits, particularly those found in India, contain high levels of contaminants and must be subjected to expensive processing prior to utilization. Alternatively, the off-gas from the boiler must be scrubbed to remove pollutants.

It is known to treat coal in a gasification process to convert the coal to a useful hydrocarbon product. For example, U.S. Patent No. 3,700,584 Johanson et. al. is directed to the hydrogenation of low rank coal having a high oxygen content.

The prior art is generally directed to the direct conversion of the carbon in the coal to a hydrocarbon product, such as methane gas. Such processes have not proved to be economically and ecologically viable means for extracting energy value from coal.

5

### SUMMARY OF THE INVENTION

One aspect of the present invention is directed to a method for the conversion of a coal-containing feedstock to a gas product including methane. The method includes the steps of contacting the coal feedstock with a treatment gas including hydrogen, such as at least about 40 weight percent hydrogen, at a reaction temperature and for a reaction time sufficient to convert at least about 90 percent of the volatile matter in the coal-containing feedstock to methane and form a purified carbon product.

According to another aspect of the present invention, a method for the treatment of a coal-containing feedstock is provided. The method includes forming a  $H_2/CO$  treatment gas by the partial oxidation of carbon and contacting the treatment gas with a coal feedstock to convert at least about 90 percent of the volatile matter in the coal-containing feedstock to methane. A purified carbon product can be recovered from the contacting step and recycled to produce the treatment gas.

According to another aspect of the present invention, a method for the treatment of coal is provided. The method includes the steps of forming a treatment gas by the partial oxidation of carbon and contacting the treatment gas with a coal feedstock at a reaction temperature and for reaction time sufficient to remove at least about 95 percent of the volatile matter in the coal-containing feedstock. The purified carbon product formed in the contacting step may be combusted in a conventional boiler.

### DESCRIPTION OF THE DRAWINGS

30

Fig. 1 illustrates a method for producing hydrogen gas for the treatment of coal according to an embodiment of the present invention.

Fig. 2 illustrates a method for producing hydrogen gas for the treatment of coal according an embodiment of the present invention.

Fig. 3 illustrates a method for producing hydrogen gas for the treatment of coal according to an embodiment of the present invention.

Fig. 4 illustrates a method for producing hydrogen gas for the treatment of coal according an embodiment of the present invention.

Fig. 5 illustrates a method for the treatment of coal using hydrogen gas according to an embodiment of the present invention.

5 Fig. 6 illustrates a method for the treatment of coal using hydrogen gas according to an embodiment of the present invention.

Fig. 7 illustrates a method for producing a hydrogen-containing gas stream for the treatment of coal according to an embodiment of the present invention.

10 Fig. 8 illustrates a method for the treatment of coal with a hydrogen-containing gas according to an embodiment of the present invention.

### DESCRIPTION OF THE INVENTION

15 The present invention is directed to the treatment of coal to produce energy value from the coal while minimizing the discharge into the atmosphere of harmful by-products that are typically associated with the conversion of coal to energy. The method enables the extraction of the available energy content from the coal in the form of high Btu gas while reducing the atmospheric discharge of harmful by-products.

20 Coal exists in relative abundance in the United States and many other regions, including third-world regions with under-developed energy production infrastructure. One of the problems associated with the use of coal is that the coal must typically be cleaned or purged of its mineral matter content, including sulfur; notwithstanding, there usually is an additional need for post-combustion gas scrubbing to meet environmental standards.

25 It is an advantage of the present invention that the coal feedstock can include low-grade coal including high-sulfur coal and other low-grade coals. Such low-grade coals are readily available and are available at low cost. As used herein, the term *low-grade coal* refers to coal having a sulfur content in excess of 2 weight percent and an ash content in excess of about 10 weight percent.

30 Generally, the method of the present invention includes providing the coal feedstock to a hydrogenation unit and contacting the coal feedstock with a treatment gas that includes  $H_2$ . The hydrocarbons in the coal that are volatile will methanate to form  $CH_4$ . When the treatment gas also includes carbon monoxide ( $CO$ ), the  $CO$  travels through the hydrogenation unit substantially unaffected. The exit gases from the hydrogenation unit ( $CH_4$ ,  $CO$  and acid gases) are

scrubbed to produce a clean, high Btu gas product. A portion of this gas product can be burned in a combined cycle generator taking advantage of the generator's high thermal-to-electric conversion efficiency. Another portion of the gas product can be combusted in a conventional boiler with the non-volatile carbon that is formed in the hydrogenation unit. The carbon will be essentially free of sulfur because the hydrogen treatment gas is a strong desulfurizing agent. In one embodiment, another portion of the carbon that is produced in the hydrogenation unit can be recycled back to the production of a  $H_2/CO$  treatment gas.

One of the important aspects of the method according to the present invention is the in-situ manufacture of large quantities of  $H_2$  at a relatively low cost. It is believed that one of the primary hindrances to the methods disclosed in the prior art for the treatment of coal is the need for high volumes of hydrogen gas and the high cost associated with the hydrogen gas. According to the present invention, high volumes of hydrogen gas can be economically generated in-situ.

Preferably, a hydrogen gas stream is produced by the reduction of steam ( $H_2O$ ) with a metal, thereby forming hydrogen gas ( $H_2$ ) and a metal oxide ( $MeO$ ). In a particularly preferred embodiment, a metal oxide is simultaneously reduced in a separate reactor to form a metal ( $Me$ ) that can subsequently be used for the steam reduction.

Referring now to the Figures, the present invention will be described in detail. Figs. 1 and 2 illustrate one embodiment of the method of the present invention using iron and iron oxide to generate hydrogen gas for subsequent hydrogenation of a coal feedstock. Figs. 1 and 2 illustrate the same apparatus with alternating flow of gaseous reactants and products to continuously form both iron and iron oxide reaction products for the process. Although this embodiment of the present invention is described with respect to oxidation/reduction reactions involving iron metal, it may be desirable to utilize metal alloys such as Fe/Ni or Fe/Cr.

As is illustrated in Fig. 1, steam (gaseous  $H_2O$ ) is generated from water **100** in a reactor **102**, such as a conventional boiler. The steam preferably exits the steam reactor **102** and is conveyed directly to a first reactor **110** through valve **114**. The temperature of the steam can be adjusted to control the temperature in the first reactor **110**.

A reducing gas, preferably a reducing gas that includes carbon monoxide ( $CO$ ), is generated in a reduction gas reactor **104**. In the embodiment illustrated

in Fig. 1, supplying carbon **106** and oxygen **108** to the reactor **104** initially generates CO from carbon and oxygen (O<sub>2</sub>). As the reaction progresses, the CO is produced from CO<sub>2</sub> by the Boudouard reaction:



The source of the carbon **106** can be, for example, coal, oil, biomass or a similar carbonaceous material. The reducing gas exiting reactor **104** is preferably at least about 90 weight percent CO and more preferably is at least  
10 about 95 weight percent CO.

The steam and the reducing gas are then supplied to first and second reactors **110** and **112**. The first and second reactors **110** and **112** can be adapted to hold molten metal or molten metal oxide through which reactant gases are passed or the reactors can be fluidized bed reactors adapted to  
15 contact the gases with particulate metal and metal oxide in a fluidized state. For the use of iron and iron oxide, the first reactor **110** and second reactor **112** are preferably fluidized beds.

The steam reactor **102** has associated valves **114** and **116** that can be switched to provide the steam to either the first reactor **110** or the second reactor  
20 **112**. In the embodiment illustrated in Fig. 1, the valve **114** is open to provide steam to the first reactor **110**. The first reactor is initially provided with iron, preferably in particulate form. In the first reactor **110** the steam reacts with the iron to form iron oxide, preferably FeO, and hydrogen gas, in accordance with the reaction:

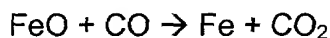


In order to maximize hydrogen production, it is preferable to feed a stoichiometric excess of H<sub>2</sub>O to the first reactor **110**. Hydrogen gas and the water vapor are  
30 removed through a valve **118** to a condenser **120** where water **122** is removed from the gas stream and pure hydrogen gas **124** is recovered.

Simultaneously, the reduction gas reactor **104** produces CO reducing gas. The valves **126** and **128** are controlled to provide the reducing gas composition to the appropriate reactor. In the embodiment illustrated in Fig. 1, the valve **128**  
35 is opened to supply reducing gas to the second reactor **112**. Excess CO **136** is preferably removed to remove oxygen from the system corresponding to the

amount of hydrogen being removed from the system. This excess CO **136** can be used as process heat, such as to heat the boiler **102**, advantageously conserving heat value in the process and maximizing the use of unreacted CO.

5 In the second reactor **112**, iron oxide is initially provided and the reducing gas composition is reacted with the iron oxide to form iron and carbon dioxide, in accordance with the reaction:



10 The iron oxide is preferably in fluidized particulate form to enable the rapid and economical formation of the iron. Advantageously, the carbon dioxide can be recycled back to the reactor **104** through valve **130** for the production of additional reducing gas. In one embodiment, sufficient CO<sub>2</sub> is recycled back to the reactor **104** such that the amount of fresh oxygen **108** supplied to reducing  
15 gas reactor **104** is only enough to maintain the desired reaction temperature. Preferably, the reaction temperature in the reduction gas reactor **104** is from about 800°C to about 1300°C.

Fig. 2 illustrates the identical apparatus as is illustrated in Fig. 1. However, in Fig. 2, the valves **114**, **116**, **118**, **126**, **128**, **130**, **132** and **134** are  
20 switched so that the first reactor **110** is the metal reduction reactor and the second reactor **112** is the steam reduction reactor. Hydrogen gas is therefore extracted from the second reactor **112** through valve **134**.

In accordance with the foregoing, it is apparent that the first and second reactors **110** and **112**, at any given point in time during the process, will include  
25 some mixture of iron and iron oxide. In one embodiment, the composition of the reactor feed in the first and second reactors is monitored and the flow of gaseous reactants is switched accordingly. Although temperature adjustments to the reactors can be made on a real-time basis, it is an advantage of the present invention that the reactors are maintained at a substantially constant reaction  
30 temperature regardless of whether the reactor is being utilized for the reduction of steam or the reduction of iron oxide. Therefore, no heating, cooling and reheating of the non-gaseous reactants is necessary. In a preferred embodiment, the reaction temperature of the first and second reactors is maintained at not greater than about 1000°C, such as from about 700°C to about  
35 900°C.

It is also an advantage of the present invention that the non-gaseous reactants (e.g., the iron and the iron oxide) are not physically moved from one location to another, such as from one reactor or reactor zone to another. On a commercially useful scale, the amount of iron and/or iron oxide in each reactor can be several hundred tons. Eliminating the need to move such a large mass of material substantially reduces the cost associated with producing the hydrogen gas. It will be appreciated that it may be desirable to supplement the iron and/or iron oxide with fresh feed due to any inherent system losses, although such supplementation should be minimal.

Other metal/metal oxide systems can be used for the production of hydrogen gas and subsequent hydrogenation of a coal feedstock according to the present invention. In one particularly preferred embodiment, tin (Sn) and tin oxide ( $\text{SnO}_2$ ) are used to form hydrogen gas.

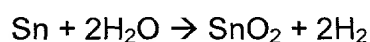
Figs. 3 and 4 illustrate an embodiment of the method of the present invention wherein hydrogen is formed using tin (Sn) and tin oxide ( $\text{SnO}_2$ ). Similar to Figs. 1 and 2, Figs. 3 and 4 illustrate the same apparatus with alternating flow of gaseous reactants and products to continuously form both metal and metal oxide reaction products.

As is illustrated in Fig. 3, steam is generated from water **300** in a steam reactor **302**, such as a conventional boiler. The steam exits the steam reactor and is conveyed to the first reactor **310** through valve **314**. The temperature of the steam can be used to partially control the reaction temperature in the first reactor **310**.

A reducing gas, preferably a reducing gas that includes carbon monoxide (CO), is generated in a reactor **304**. In the embodiment illustrated in Fig. 3, supplying carbon **306** and oxygen **308** to the reactor initially generates CO from carbon and oxygen ( $\text{O}_2$ ). As the reaction progresses, the CO is produced from  $\text{CO}_2$  by the Boudouard reaction. The source of the carbon **306** can be, for example, coal or a similar carbonaceous material. The reducing gas exiting the reactor **304** is preferably at least about 90 weight percent CO and more preferably is at least about 95 weight percent CO.

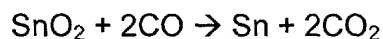
The steam and the reducing gas are then supplied to first and second reactors **310** and **312**. The steam is preferably reacted with molten tin metal by passing the gaseous steam through a pool of the molten tin metal. The molten tin metal will oxidize, forming  $\text{SnO}_2$  particulates dispersed in the molten Sn metal.

Therefore, the first reactor **310** and second reactor **312** are preferably large reactors adapted to heat and contain molten tin and a mixture (slurry) of molten tin with tin oxide. The steam reactor **302** has associated valves **314** and **316** that can be switched to provide the steam to either the first reactor **310** or the second reactor **312**. In the embodiment illustrated in Fig. 3, the valve **314** is open to provide steam to the first reactor **310**. The first reactor **310** is initially provided with a pool of molten tin having a temperature of at least about 232°C (the melting point of tin) and preferably from about 600°C to about 800°C. In the first reactor **310** the steam reacts with the molten tin to form SnO<sub>2</sub> and hydrogen gas, in accordance with the reaction:



To maximize hydrogen production, it is preferable to feed a stoichiometric excess of H<sub>2</sub>O to the first reactor **310**. Hydrogen gas and the excess water vapor are removed through a valve **318** to a condenser **320** where water **322** is removed from the gas stream and pure hydrogen gas **324** is recovered.

Simultaneously, the reduction gas reactor **304** produces CO reducing gas by the Boudouard reaction. The valves **326** and **328** are controlled to provide the reducing gas composition to the appropriate reactor. In the embodiment illustrated in Fig. 3, the valve **328** is opened to supply reducing gas to the second reactor **312**. Excess CO **336** is preferably removed to remove oxygen from the system corresponding to the amount of hydrogen being removed from the system. This excess CO **336** can be used as process heat, such as to heat the boiler **302**. In the second reactor **312**, tin oxide is initially provided and the reducing gas composition is reacted with the tin oxide to form molten tin metal and carbon dioxide, in accordance with the reaction:



The tin oxide is preferably in particulate form to enable the rapid and economical reduction of the tin oxide to tin metal. The tin oxide can initially be provided as a slurry of the oxide in molten tin metal. Advantageously, the carbon dioxide can be recycled back to the reactor **304** through valve **328** for the production of additional reducing gas. In one embodiment, sufficient CO<sub>2</sub> is recycled back to the reactor **304** such that the amount of fresh oxygen **308** supplied to Boudouard



reactor **304** is only enough to maintain the desired reaction temperature, which is preferably from about 800°C to about 1300°C. This also maximizes the use of unreacted CO.

Fig. 4 illustrates the identical apparatus as is illustrated in Fig. 3. However, in Fig. 4, the valves **314**, **316**, **318**, **326**, **328**, **330**, **332** and **334** are switched so that the first reactor **310** is the metal reduction reactor and the second reactor **312** is the steam reduction reactor. Hydrogen gas is therefore extracted from the second reactor **312** through valve **334**.

It is apparent that the reactors **310** and **312**, at any given point in time during the process, will include some mixture of tin and tin oxide. In one embodiment, the composition of the reactor feed in the first and second reactors is monitored and the flow of gaseous reactants is switched accordingly. Although temperature adjustments to the reactors can be made on a real-time basis, it is an advantage of the present invention that the reactors are maintained at a substantially constant reaction temperature regardless of whether the reactor is being utilized for the reduction or the oxidation of the tin/tin oxide. Therefore, no heating, cooling and reheating of the non-gaseous reactants is necessary. In a preferred embodiment for the production of hydrogen from Sn/SnO<sub>2</sub>, the reaction temperature of the first and second reactors is maintained at a temperature of at least about 232°C and not greater than about 1120°C (the melting point of SnO<sub>2</sub>), such as from about 400°C to about 800°C. Lower reaction temperatures are preferred to minimize large SnO<sub>2</sub> particles whereas higher temperatures increase reaction kinetics.

As with the iron/iron oxide system described hereinabove, it is also an advantage of the present invention that the non-gaseous reactants (i.e., the molten tin and the tin oxide) are not physically moved from one location to another, such as from one reactor or reactor zone to another. On a commercially useful scale, the amount of molten tin and/or tin oxide in each reactor can be several hundred tons. By eliminating the need to move such a large mass of material, the cost associated with producing the hydrogen gas is substantially reduced. It will be appreciated that it may be desirable to supplement the tin and/or tin oxide with fresh feed due to any inherent system losses, although such supplementation should be minimal.

The hydrogen gas stream that is produced according to the foregoing method has a high purity and preferably includes greater than about 99 weight percent hydrogen and more preferably greater than about 99.9 weight percent

hydrogen after removal of residual water in the condenser. It is an advantage of this embodiment of the invention that the hydrogen gas does not require separation from another gas species such as carbon monoxide (CO).

5 The substantially pure hydrogen gas stream that has been formed by the reduction of steam can then be used as the treatment gas for the treatment of coal in accordance with the present invention. The integration of the foregoing method and apparatus with a coal treatment method and apparatus will be more clearly understood with reference to Figs. 5 and 6

10 Referring to Fig. 5, water is heated in a boiler **502** to produce steam. The path of the gaseous steam is controlled by valve **514** and is directed to a reactor **510** where a metal (e.g., tin) is oxidized to a metal oxide (e.g., tin oxide), thereby forming hydrogen gas in accordance with the method described with respect to Figs. 1-4. The hydrogen gas stream is routed through valve **518** and includes some excess water that can be removed in a condenser **520**. A heat exchange device **532** can be utilized to extract heat value from the hydrogen and steam gas stream and the extracted heat can be used as process heat, such as to heat the boiler **502**.

15 Simultaneously, carbon and oxygen are supplied to a reactor **504** where CO is generated and supplied to reactor **512** through valve **526**. In the reactor **512**, the metal oxide is reduced to the metal by reacting the metal oxide with the CO. It will be appreciated with reference to Figs. 1-4 that CO<sub>2</sub> exiting the reactor **512** through valve **516** can be routed to the reactor **504** to generate CO.

20 The hydrogen-containing gas stream is conveyed to a hydrogenation unit **540** where the hydrogen gas is contacted with the coal feedstock. The hydrogenation unit **540** can be a fluidized bed reactor or other reactor that is suitable for the treatment of particulate coal. The unit is preferably operated at ambient or near ambient pressure, such as a pressure of not greater than about 5 psi. It is an advantage of the present invention that the hydrogenation unit **540** is not operated at a substantial elevated pressure in order to treat the coal and produce substantial quantities of methane-containing product gas.

25 The reaction in the hydrogenation unit **540** is preferably carried out at a temperature of at least about 700°C and preferably not greater than about 1100°C, such as from about 800°C to about 900°C. The reaction that occurs in the hydrogenation unit is exothermic and therefore the need for the addition of external heat is minimal.

Referring to Fig. 6, the same apparatus as in Fig. 5 is illustrated wherein the valves **514**, **516**, **518** and **526** are switched such that a hydrogen gas stream is produced by steam oxidation of iron in the reactor **512** and is routed to the hydrogenation unit **540**. Thus, a substantially constant high volume flow of hydrogen gas can be supplied to hydrogenate the incoming coal feedstock.

Coal is a complex mixture of chemical compounds that are primarily organic compounds. While coal is predominately a hydrocarbon, impurities such as sulfur and nitrogen are trapped in the coal. These impurities lead to the formation of sulfuric acid and nitric acid if released into atmosphere and must be scrubbed from the off-gas in a typical coal-based power plant. Further, when the coal is burned, the carbon combines with oxygen in the atmosphere and forms CO<sub>2</sub>, a well-known greenhouse gas that traps the earth's heat.

According to the present invention, when the coal feedstock is treated in the hydrogenation unit **540** with the hydrogen treatment gas, the coal is not burned. Rather, it is heated to remove the volatile components in the coal. These volatile components advantageously form a high Btu product gas containing methane. The residue from the removal of the volatiles is purified by the action of the hydrogen, including removal of sulfur and nitrogen, and the resulting purified carbon (i.e., coke) can be combusted in a conventional boiler with methane and without the formation of detrimental by-products.

The methane-containing gas stream that is produced by the hydrogenation reaction can be treated in a scrubber **542** to remove contaminants from the gas stream. For example, the scrubber can apply a caustic solution to form a benign waste stream including, for example, chloride compounds. Sulfur in the coal will form hydrogen sulfide (H<sub>2</sub>S) that can be scrubbed from the methane gas stream. It is advantageous to scrub the methane gas stream prior to combustion, since combustion creates a higher volume gas stream.

The methane gas can be combusted on-site to generate electricity, or can be treated to remove CO (if any) and provided to end-users as a pipeline gas. Further, a portion of the CH<sub>4</sub> can be cycled back to the other unit operations to provide process heat. For example, a portion of the CH<sub>4</sub> can be diverted to the reactors **510** and **512** and/or can be used to power the boiler **502**.

The hydrogenation treatment advantageously removes the impurities in the coal. The treated carbon exiting the hydrogenation unit will be a clean coke product that includes some ash and preferably includes not greater than about 0.1 weight percent sulfur.

According to one embodiment of the present invention, a portion of the coke can be cycled back to the reactor **504** for the production of CO. In the reactor **504** the carbon will be converted to CO and the ash can be easily removed. The remaining coke from the hydrogenation unit **540** can be burned in a conventional boiler with methane for additional energy production. Due to the purity of the incoming treated carbon, the ash by-product from the boiler advantageously has a low carbon content and is a valuable commodity. For example, the ash can be readily used as a filler in cementitious compositions.

Further, a portion of the H<sub>2</sub> gas that is produced by the reduction of steam can be diverted from the hydrogenation unit **540** and used directly as a fuel source, either alone or in combination with the methane gas. For example, the hydrogen can be directly fired in a boiler.

The treatment of coal according to an alternative embodiment of the present invention can be carried out using a treatment gas that includes both hydrogen and carbon monoxide (CO).

A flowsheet illustrating the means for generating a H<sub>2</sub>/CO gas stream by partial oxidation of carbon is illustrated in Fig. 7. Water is provided to a boiler **704** to generate steam. The steam is combined with carbon in a reactor **710** at an elevated temperature, such as from about 900°C to about 1300°C, to form H<sub>2</sub> and CO which exits the reactor **710** with excess H<sub>2</sub>O. This gas stream can be passed through a heat exchanger **712** to conserve process heat and then through a condenser **714** to remove the excess water. The dried gas stream, which is a mixture of H<sub>2</sub> and CO in a 1:1 molar ratio, can then be used as a treatment gas in a hydrogenation unit.

Referring to Fig. 8, a reduction gas stream is provided including H<sub>2</sub> and CO, such as from the partial oxidation of carbon illustrated in Fig. 7. Specifically, water is heated in a boiler **804** to form steam, which is contacted with carbon in reactor **810** to form H<sub>2</sub> and CO. After removal of excess water in a condenser **814**, the treatment gas is provided directly to the hydrogenation unit **840**. Simultaneously, a coal feedstock is provided to the hydrogenation unit **840** to be reacted with the treatment gas. The coal feedstock can optionally be subjected to pre-treatment, such as to comminute the coal to a maximum particle size of about 1mm. Preferably, the reaction takes place at a temperature of at least about 600°C and not greater than about 1100°C, such as from about 700°C to about 900°C.

The CO essentially passes through the hydrogenation unit **840** without reacting with the feedstock. The hydrogen reacts with the hydrocarbon species in the coal to form methane. The methane gas stream can be scrubbed in scrubber **842** to remove contaminants, as is discussed above with respect to Figs. 5 and 6.

In the embodiment illustrated in Fig. 8, the resulting scrubbed product gas stream is split to provide power generation in a generator **860** and to cycle a portion of the methane-containing product gas to a conventional boiler **862** to generate process heat and/or additional electricity. Preferably, a portion of the scrubbed product gas stream is combined in the boiler **862** with the purified carbon product. Due to the use of very clean carbon and methane gas, the boiler can operate at sufficiently low temperatures to preclude formation of nitrogen oxides while maintaining high conversion efficiencies of the carbon.

It should also be noted with respect to Fig. 8 that a portion of the H<sub>2</sub>/CO treatment gas can be diverted prior to the hydrogenation unit **840** and mixed with the methane-containing gas.

The generator **860** can advantageously be a combined cycle generator. In a combined cycle generator, exhaust heat from a first system, referred to as the top cycle, is used to generate power in a second system, referred to as the bottom cycle. Such combined cycle systems typically employ a combustion turbine in the top cycle, and a steam turbine in the bottom cycle. A heat recovery steam generator uses the hot exhaust gas from the combustion turbine to produce steam that drives one or more steam turbines that can be used to generate electricity. Combined cycle generators can operate with up to about 55 percent efficiency, whereas traditional boilers operate at only about 37 percent efficiency.

As is discussed above, the primary feedstock to the hydrogenation unit of the present invention can be a low-grade coal as well as a high-grade coal. Indeed, low-grade coal feedstock can be advantageous since they are generally available at a low cost. In one embodiment, the coal feedstock is a low grade coal having at least 1 weight percent sulfur, such as at least about 2 weight percent sulfur. The coal can be pre-treated in a pre-treatment unit **864** such as to comminute the coal and remove surface moisture. For example, it is desirable to comminute the coal to reduce the maximum particle size to not greater than about 1 mm. When the feedstock is comminuted coal, the hydrogenation process results in a highly purified and finely divided carbon product.

The hydrogenation treatment according to the present invention converts the volatiles (e.g., hydrocarbons) in the coal to methane. Preferably, at least about 90 weight percent, more preferably at least about 95 weight percent and even more preferably at least about 99 weight percent of the volatile matter is converted to methane.

As is discussed above, the coal frequently includes contaminants, including high levels of sulfur. The method of the present invention advantageously takes advantage of the fuel value of the volatile matter and simultaneously strips the contaminants from the coke. The sulfur will react with the hydrogen gas and form hydrogen sulfide, which can be easily removed from the waste stream.

Further, the method of the present invention enables the production of energy from the coal feedstock while producing significantly less carbon dioxide per kilowatt-hour of energy produced than is typically produced in a conventional coal-fired power plant. The reduction in carbon dioxide emission is an important environmental factor and adds further value to the method.

In one embodiment, the carbon that is produced in the hydrogenation unit (the coke) is an activated carbon. Activated carbon is an amorphous form of carbon characterized by high adsorptivity for many gases, vapors and colloidal solids. The internal surface area of the activated carbon exceeds several hundred  $\text{m}^2/\text{gram}$  and the density is not greater than about  $0.85 \text{ g/cm}^3$ . Activated carbon has a high value and can be used for air and water purification, waste treatment, removal of mercury (Hg) and  $\text{SO}_x$  from stack gases and the like. Thus, the hydrogenation process when applied to coal provides a high Btu value gas stream as well as a valuable by-product.

While various embodiments of the present invention have been described in detail, it is apparent that modifications and adaptations of those embodiments will occur to those skilled in the art. However, it is to be expressly understood that such modifications and adaptations are within the spirit and scope of the present invention.